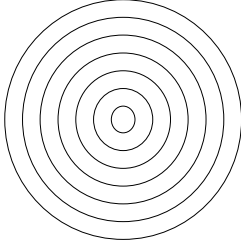
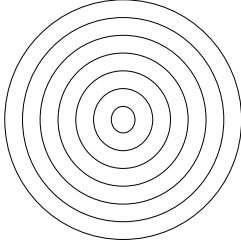
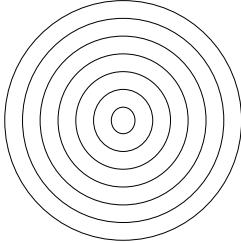
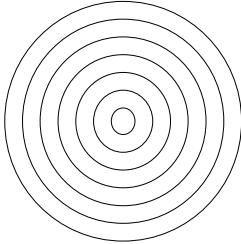


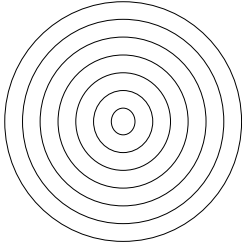
# ELECTRON CONFIGURATION PRACTICE

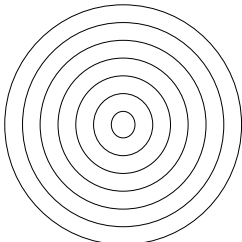
1. Element:	A) # of Electrons:	D) Bohr Diagram: (some shells may not be used) 
B) Electron configuration (use multiple lines if necessary): _____		
C) Electrons in each energy level: 1 <sup>st</sup> ____ 2 <sup>nd</sup> ____ 3 <sup>rd</sup> ____ 4 <sup>th</sup> ____ 5 <sup>th</sup> ____ 6 <sup>th</sup> ____ 7 <sup>th</sup> ____	F) Lewis Dot Diagram:	
E) Number of valence electrons:		

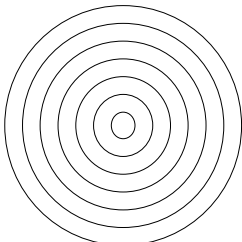
2. Element:	A) # of Electrons:	D) Bohr Diagram: (some shells may not be used) 
B) Electron configuration (use multiple lines if necessary): _____		
C) Electrons in each energy level: 1 <sup>st</sup> ____ 2 <sup>nd</sup> ____ 3 <sup>rd</sup> ____ 4 <sup>th</sup> ____ 5 <sup>th</sup> ____ 6 <sup>th</sup> ____ 7 <sup>th</sup> ____	F) Lewis Dot Diagram:	
E) Number of valence electrons:		

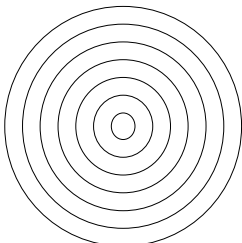
3. Element:	A) # of Electrons:	D) Bohr Diagram: (some shells may not be used) 
B) Electron configuration (use multiple lines if necessary): _____		
C) Electrons in each energy level: 1 <sup>st</sup> ____ 2 <sup>nd</sup> ____ 3 <sup>rd</sup> ____ 4 <sup>th</sup> ____ 5 <sup>th</sup> ____ 6 <sup>th</sup> ____ 7 <sup>th</sup> ____	F) Lewis Dot Diagram:	
E) Number of valence electrons:		

4. Element:	A) # of Electrons:	D) Bohr Diagram: (some shells may not be used) 
B) Electron configuration (use multiple lines if necessary): _____		
C) Electrons in each energy level: 1 <sup>st</sup> ____ 2 <sup>nd</sup> ____ 3 <sup>rd</sup> ____ 4 <sup>th</sup> ____ 5 <sup>th</sup> ____ 6 <sup>th</sup> ____ 7 <sup>th</sup> ____	F) Lewis Dot Diagram:	
E) Number of valence electrons:		

<b>5. Element:</b>	<b>A) # of Electrons:</b>	<b>D) Bohr Diagram:</b> (some shells may not be used) 
<b>B) Electron configuration (use multiple lines if necessary):</b> _____		
<b>C) Electrons in each energy level:</b> 1 <sup>st</sup> _____ 2 <sup>nd</sup> _____ 3 <sup>rd</sup> _____ 4 <sup>th</sup> _____ 5 <sup>th</sup> _____ 6 <sup>th</sup> _____ 7 <sup>th</sup> _____	<b>F) Lewis Dot Diagram:</b>	
<b>E) Number of valence electrons:</b>		

<b>6. Element:</b>	<b>A) # of Electrons:</b>	<b>D) Bohr Diagram:</b> (some shells may not be used) 
<b>B) Electron configuration (use multiple lines if necessary):</b> _____		
<b>C) Electrons in each energy level:</b> 1 <sup>st</sup> _____ 2 <sup>nd</sup> _____ 3 <sup>rd</sup> _____ 4 <sup>th</sup> _____ 5 <sup>th</sup> _____ 6 <sup>th</sup> _____ 7 <sup>th</sup> _____	<b>F) Lewis Dot Diagram:</b>	
<b>E) Number of valence electrons:</b>		

<b>7. Element:</b>	<b>A) # of Electrons:</b>	<b>D) Bohr Diagram:</b> (some shells may not be used) 
<b>B) Electron configuration (use multiple lines if necessary):</b> _____		
<b>C) Electrons in each energy level:</b> 1 <sup>st</sup> _____ 2 <sup>nd</sup> _____ 3 <sup>rd</sup> _____ 4 <sup>th</sup> _____ 5 <sup>th</sup> _____ 6 <sup>th</sup> _____ 7 <sup>th</sup> _____	<b>F) Lewis Dot Diagram:</b>	
<b>E) Number of valence electrons:</b>		

<b>8. Element:</b>	<b>A) # of Electrons:</b>	<b>D) Bohr Diagram:</b> (some shells may not be used) 
<b>B) Electron configuration (use multiple lines if necessary):</b> _____		
<b>C) Electrons in each energy level:</b> 1 <sup>st</sup> _____ 2 <sup>nd</sup> _____ 3 <sup>rd</sup> _____ 4 <sup>th</sup> _____ 5 <sup>th</sup> _____ 6 <sup>th</sup> _____ 7 <sup>th</sup> _____	<b>F) Lewis Dot Diagram:</b>	
<b>E) Number of valence electrons:</b>		